J. Nano- Electron. Phys. 3 (2011) No1, P. 507-513 © 2011 SumDU (Sumy State University)

PACS numbers: 68.37. Hk, 81.15.Ef, 68.55.aq, 78.66. - w, 61.05.C -

GROWTH STRUCTURAL AND OPTICAL PROPERTIES OF THE THERMALLY EVAPORATED TIN DISELENIDE (SnSe₂) THIN FILMS

R. Sachdeva¹, M. Sharma¹, A. Devi¹, U. Parihar¹, N. Kumar¹, N. Padha^{1*}, C.J. Panchal²

- ¹ University of Jammu, Baba Sahib Ambedker Road, 180006, Jammu Tawi, India E-mail: nareshpadha@yahoo.com
- ² M. S. University of Baroda 390001, Vadodara, India

Tin diselenide $(SnSe_2)$ compound was prepared by melt-quenching technique from its constituent elements. The phase structure and composition of the chemical constituents present in the bulk has been determined using X-ray diffraction (XRD) and energy dispersion X-ray analysis (EDAX) respectively. $SnSe_2$ thin films were grown using direct thermal evaporation of $SnSe_2$ compound material on chemically cleaned glass substrate, which were held at different substrate temperatures. X-ray diffraction and Scanning Electron Microscopy (SEM) were used to examine structure and surface morphology of the films. The investigations show that the films grown above 150 °C are single phase and polycrystalline in nature. VIS-IR Spectra of the films were recorded in the wavelength range 380 nm to 1300 nm. The data has been analyzed to find optical parameters like absorption coefficient and energy bandgap.

Keywords: SnSe₂, THIN FILMS, THERMAL EVAPORATION, SEM, XRD, TRAMS-MITTANCE MEASUREMENTS.

(Received 04 February 2011, in final form 19 June 2011)

1. INTRODUCTION

The science and technology of thin film materials have an essential role in the high-tech industries. While until recently the major exploitation of thin films has been in the field of microelectronics, today there are numerous and growing applications in communications, optical electronics, and coatings of all kinds and in energy generation and conversation strategies [1]. Tin diselenide has been used in the field of photovoltaic and infrared optoelectronic devices, radiation detectors, holographic recording systems, electrical switching and polarity dependent memory switching devices etc. [2, 3]. The developed methods for synthesis of tin diselenide thin films which have been described in the literature often employ more sophisticated techniques such as: reactive evaporation [2, 3], vacuum evaporation [4-6], electrode deposition [7, 8] and hot wall epitaxy [9]. In the present work, study was done to analyze the effect of substrate temperature on the structural and optical properties of $SnSe_2$ thin films.

507

2. EXPERIMENTAL DETAILS

2.1 Film Preparation

Thin films of tin diselenide were prepared from the bulk material by a thermal evaporation technique on chemically-cleaned glass substrates at different temperatures ranging from RT to 300 °C at a reduced pressure of 10^{-5} Torr (1 Torr = 1.333×10^{-2} Pa) using the conventional vacuum coating unit (Hind Hivac Co. Ltd., India). Molybdenum boat was used to evaporate SnSe₂ powder. The deposition rate of 0.4-0.6 nm/s for typical thickness of 100 nm was continuously monitored during the deposition using a quartz crystal thickness monitor DTM-101 (Hind Hi Vac., India). The substrate temperature was measured using chromel-alumel thermocouple kept in good thermal contact with the substrate.

2.2 Characterization techniques

The structural characterization of the thermally evaporated $SnSe_2$ thin films were carried out using an X-ray diffractometer (Philips X'Pert PRO MRD HR X-Ray Diffraction System) in 2θ range of 10 ° to 80 ° at a scan rate of 0.05 s⁻¹ using CuK α radiation. The morphology of the film was studied by scanning electron microscopy (FEI, Quanta 200F, Netherland). The optical transmittance measurements used to determine the absorption coefficient and the optical bandgap (E_g) were carried out using monochrometer (CM110), photodetectors (New port 818-SL and 818-SR) and lock in amplifier. The whole setup was automated using LabVIEW8.2.

3. RESULTS AND DISCUSSIONS

3.1 Structural characterization of SnSe₂ thin films

Fig. 1 a shows the X-ray diffraction pattern of the synthesized $SnSe_2$ powder. The sharp peaks present in X-ray diffraction pattern indicate that the prepared $SnSe_2$ has hexagonal structure and polycrystalline in nature [10]. EDAX analysis of the compound was done to confirm the stoichiometric. The EDAX pattern of the powder is also shown in Fig. 1 b. Table 1 presents EDAX data which indicates the stoichiometry of elements which verifies formation of $SnSe_2$ phase in synthesized powder.



Fig. 1 – X-Ray diffraction (a) and EDAX pattern (b) of $SnSe_2$ powder

508

X-ray diffraction pattern of the SnSe_2 thin films deposited at different substrate temperatures are shown in Fig. 2. From the figure, it is clear that the substrate temperature has significant influence on the quality of the films. For instance, the films grown at room temperature to 150 °C are amorphous in nature.

Table 1 – Experimental and theoretical EDAX data of the prepared $SnSe_2$ compound

Experimental			Theoretical		
	Atomic %	Weight %		Atomic %	Weight %
Sn	35.6	35	Sn	35.27	34.78
Se	64.2	65	Se	64.72	65.22



Fig. 2 – The XRD pattern of SnSe₂ films at different substrate temperatures

The initialization of the crystalline phase has started when the substrate temperature increased above 150 °C. Further, thin films grown at substrate temperature 150-300 °C are single phase having orientation of the grains in (101) plane. The crystallite size of the films deposited at different substrate temperatures were calculated using the well known Scherrer's formula [11].

$$d = \frac{0.9\lambda}{B\cos\theta} \tag{1}$$

Where λ is the wavelength of CuK α ($\lambda = 1.5418$ Å) radiation, B is the fullwidth half-maximum (FWHM). The position and FWHM of the (101) peak, along with the calculated values of the crystallite size for films grown at different substrate temperatures are shown in Table 2.

The morphological analysis of the films grown at different substrate temperatures were carried out using Scanning Electron Microscopy (SEM). Fig. 3 shows SEM images of the films grown at different substrate temperatures. It confirms the inferences of the XRD measurements that grain/crystallite size increases with substrate temperature.

Substrate Temperature $(T_s), \ ^{\circ}\mathrm{C}$	Position of peak (101), degrees	FWHM, radians	Crystallite size, nm
150	30.9069	0.44	31.20
200	30.9069	0.39	35.15
250	30.9658	0.29	46.80
300	31.0339	0.12	111.44

Table 2 – Position of (101) peak, Crystallite size and FWHM of $SnSe_2$ films grown at different temperatures



Fig. 3 – SEM images of the films grown at 150 °C (a), 200 °C (b), 250 °C (c), 300 °C (d)

3.2 Optical characterization

3.2.1 Transmission Spectra and Absorption Coefficient

The plots of transmission spectra versus wavelength and absorption coefficient versus wavelength are shown in Fig. 4.



Fig. 4 – Transmission spectra (a) and the absorption coefficient α (b) of $SnSe_2$ thin films deposited at different substrate temperatures

3.2.2 Absorption coefficient and energy bandgap

Transmission spectra in the wavelength range 380 nm to 1300 nm at normal incidence of as-deposited (at different substrate temperatures) films were recorded. From these spectral data optical absorption coefficient was calculated using Lambert's law [12],

$$\ln\left[\frac{I_0}{I}\right] = 2.303A = \alpha d \tag{2}$$

where I and I_0 are the intensity of the incident and transmitted light, A is the optical absorbance and d is the film thickness. The absorption coefficient α was found to follow the relation [13],

$$\alpha h v = B \left(h v - E_{g} \right)^{1/2}, \tag{3}$$

where B is a constant and E_g is the bandgap energy. Plot of $(\alpha h v)^2$ versus the photon energy (h v) for films deposited at different substrate temperatures is shown in Fig. 5.

Linearity of the plot indicates that the material is of direct bandgap [14]. The values of bandgap estimated from the plot of the films are shown in Table 3.

Table 3 – Variation of bandgap with substrate temperature

Substrate Temperature T_s , °C	E_g , eV
RT	1.72
50	1.72
100	1.67
150	1.67
200	1.68
250	1.65
300	1.62



Fig. 5 – Plot of images of $(\alpha hv)^2$ versus the photon energy (hv) of the SnSe₂ films

The bandgap calculated on the basis of this analysis is found to decrease with increase in the substrate temperature and vary from 1.62 eV to 1.72 eV in the manner as shown in Table 3. These values are in good agreement with the bandgap values as reported by many earlier workers [15-17]. The decrease in the direct bandgap value with increasing temperature in semiconductors was studied by Olguin et al. [18]. The authors reported two main reasons for this behavior. The first reason for the direct bandgap shift is the effect of thermal expansion of the lattice, which relates to the change of electron energies with volume. The second reason is the direct renormalization of the band energies due to temperature dependence of the electron-phonon interactions.

4. CONCLUSION

In this work, $SnSe_2$ thin films were prepared from stoichiometric compound using direct thermal evaporation method. X-ray diffraction and SEM analyses revealed that the Thin Film grain size increased with the substrate temperature and it has provided maximum value of 111.4 nm at the substrate temperature of 300 °C having preferred orientation in the (101) direction. The transmission spectrum was analyzed to obtain optical parameters. The films have absorption coefficient in the range 10 to 20 and provides direct bandgap which decreases with increase in substrate temperature. Typical value of the bandgap which has been obtained at 300 °C is 1.62 eV.

REFERENCES

- 1. M. Ohring, *The material and science of thin films* (New York, FRG: Academic press: 1992).
- 2. K.J. John, B. Pradeep, E. Mathai, J. Mater. Sci. 29, 1581 (1994).

- 3. A. Agarwal, M.N.E. Vashi, D. Lakshminarayana, N.M. Batra, J. Mater. Sci. Mater. El. 11, 67 (2000).
- 4. T. Subba Rao, A.K. Chaudhuri, J. Phys.: D Appl. Phys. 18, L35 (1985).
- 5. T. Subba Rao, A.K. Chaudhuri, J. Phys.: D Appl. Phys. 19, 861 (1986).
- 6. D.P. Padiyan, A. Marikani, K.R. Murali, Cryst. Res. Technol. 35, 949 (2000).
- R.D. Engelken, A.K. Berry, T.P. Van Qoren, J.L. Boone, A. Shahnazary, J. Electrochem. Soc. 133, 581 (1986).
- B. Subramanian, T. Mahalingam, C. Sanjeeviraja, M. Jayachandran, M.J. Chockalingam, *Thin Solid Films* 357, 119 (1999).
- 9. J.P. Singh, R.K. Bedi, J. Appl. Phys. 68, 2776 (1990).
- 10. Evans, Hazelwood, Ph. D Thesis R. Hazelwood Reading, *Private Communication*, (1970).
- 11. B.D. Cullity, *Elements of X-ray Differaction* (London, FRG: Addison Willy: 1978).
- 12. A. Adachi, A. Kudo, T. Sakata, Bull. Cem. Soc. Jpn. 68, 3283 (1995).
- 13. J.I. Pankove, *Optical processes in Semiconductors*, (Englewood Cliffs: Prentice Hall: NJ: 1971).
- N.M. Shah, J.R. Ray, K.J. Patel, V.A. Kheraj, M.S. Desai, C.J. Panchal, B. Rehani, *Thin Solid Films* 517, 3639 (2009).
- 15. G. Domingo, R.S. Itoga, C.R. Kannewurf, Phys. Rev. 143, 536 (1966).
- 16. V.P. Bhatt, K. Gireesan, C.F. Desai, Cryst. Res. Technol. 25, 209 (1990).
- 17. M.M. El-Nahass, J. Mater. Sci. 27, 6597 (1992).
- 18. D. Olguin, M. Cardona, A. Cantrarero, Solid State Commun. 122, 575 (2002).